

SESQUITERPENE LACTONES AND OTHER CONSTITUENTS FROM *EREMANTHUS SEIDELII*, *E. GOYAZENSIS* AND *VANILLOSMOPSIS* *ERYTHROPAPPA*

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Key Word **Index**-*Eremanthus seidelii*; *Eremanthus goyazensis*, *Vanillosmopsis erythropappa*; Compositae; Vernonieae; sesquiterpene lactones; 3,10-epoxygermacranolides; 4,5-dihydroeremantholides; X-ray analysis.

Abstract-Extraction of the aerial parts of *Eremanthus seidelii* and reinvestigation of *E. goyazensis* gave four new 4,5-dihydrofuranoheliangolides and two new 4,5-dihydroeremantholides as well as some known sesquiterpene lactones. X-Ray analysis of one of the 4,5-dihydroeremantholides showed that the new lactones from *E. seidelii* and *E. goyazensis* differ in C-4 stereochemistry from isomers previously isolated from *Eremanthus* and related species. Criteria based on NMR spectroscopy for distinguishing between the two series are given. The C-4 stereochemistry of zexbrevin is discussed and the tentative conclusion is reached that in the earlier correlation between tagitinin A and zexbrevin C-4 epimerization took place at the oxidation stage. Reinvestigation of *Vanillosmopsis erythropappa* wood resulted in isolation of several lactones not previously reported from this species.

INTRODUCTION

Extracts of various *Eremanthus* species (Compositae, tribe Vernonieae) possess schistosomicidal properties [1-3]. The active constituent in the heartwood oil of *E. elaeagnus* is the guaianolide eremanthin (1) [1,3,4] which is also found in the schistosomicidal wood oil of *Vanillosmopsis erythropappa* [1,4,5], in the aerial parts of other members of subtribe Lychnophorinae [6-14] and in a few *Vernonia* species [15,16]. The herbaceous parts of *E. goyazensis* yield the schistosomicidal and cytotoxic heliangolide goyazensolide (4s) [17,18] which is closely related to the cytotoxic lactones eremantholides A, B and C (5a-c) from the wood of *E. elaeagnus* [19,20]. Analogues of goyazensolide and the eremantholides have since been reported from other Lychnophorinae [6-14,21-27].

In order to secure more eremanthin and goyazensolide for further evaluation as schistosomicides we have extracted collections from Minas Gerais State, Brazil, of *E. goyazensis* and what was initially thought to be *E. elaeagnus* but what on closer study turned out to be the recently described *E. seidelii* MacLeish and Schumacher [28]. ‡ Unlike *E. elaeagnus* the aerial parts of *E. seidelii* did not furnish eremanthin, but gave the new sesquiterpene lactones 2a-d, 3a and 3d as well as the flavone isorhamnetin (6). Substances isolated from the aerial

parts of *E. goyazensis* were, in addition to goyazensolide (4a) and its 11,13-dihydro derivative, the sesquiterpene lactones 2a-c, 3a, 3b, 5c (eremantholide C) [20], 5d [9,27] and isorhamnetin. Finally, extraction of *Vanillosmopsis erythropappa* wood with ethyl acetate after preliminary extraction with hexane to remove the wood oil which contains eremanthin and other sesquiterpene lactones [1,4,8,29] gave goyazensolide (4a), 15-deoxygoyazensolide (4b) [18,30], lychnopholide (4c) [12], costunolide (7) and bisabolol (8). Compounds 4b, 7 and 8 have been isolated previously from other parts of *V. erythropappa* [1,29-31].

RESULTS AND DISCUSSION

We begin by considering lactones 2a-d whose mass, ¹H and ¹³C NMR spectra (Tables 1 and 2) showed that they were 1-oxo-3,10-epoxy-8-acyloxygermacra-2,11(13)-dien-6,12-olides differing from each other only in the nature of the acyloxy group on C-8. Extensive decoupling in the usual way which will not be detailed here established the sequence C-4 through C-9 and the attachment of the α,β-unsaturated lactone ring to C-7. Closure of the lactone to C-6 was assumed on the basis of the relative shifts of H-6 and H-8 and by analogy with lactone constituents of other *Eremanthus* species. The nature of the various acyloxy groups was also evident from the spectroscopic evidence; interestingly acetylation of 2c furnished not the expected 2d, but the isomeric monoacetate 2e as demonstrated by the ¹H NMR shifts on addition of trichloroacetylisocyanate (TAI) (Table 1).

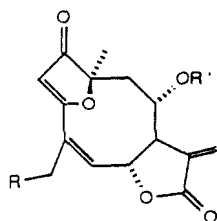
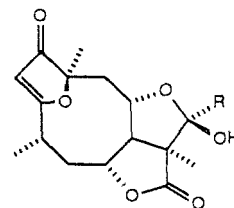
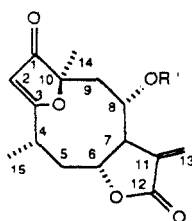
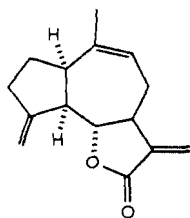
As to stereochemistry, α-orientation of the various acyloxy substituents at C-8 could be deduced from both the chemical shift of H-8 near 64.4 and the coupling constants involving H-7, H-8 and H-9a,b (*J*_{7,8} ~ 3, *J*_{8,9a}

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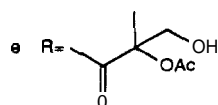
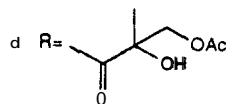
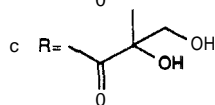
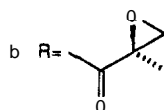
‡ According to these authors, *E. seidelii* is restricted to the cerrado surrounding the Furnas reservoir in southwestern Minas Gerais where our material was collected. It is closely related to *E. elaeagnus* but distinguishable by the number of heads per capitulescence, pappus colour, leaf shape and flowering period.

= 12, $J_{8,9b}$ -2 Hz); for comparison ^1H and ^{13}C NMR spectra of isomer **10a** (zexbrevin) with a B-orientated C-8 substituent (H-8's near 65.3, $J_{7,8}$'s ~ 1, $J_{8,9a}$'s ~ 5, $J_{8,9b}$ -2-3 Hz) are included in Tables 1 and 2 (for further discussion of zexbrevin stereochemistry and its implications see below). Consequently **2a** was 4,5-dihydro-15-desoxygoyazensolide (**2a** or **9a**).

Formulae of two lactones obtained as a mixture from *Eremanthus bicolor* and originally allotted structures **2a** and **2f** [7], and of the corresponding angelate from *Piptolepis ericoides* presumed to be **2g**[8], have recently been altered to **9a**, **9b** and **9c** as a result of NOE experiments on related compounds [32]. ¹H NMR data for these substances taken from the literature are included

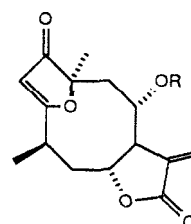
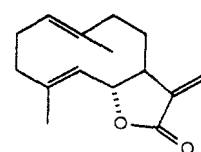
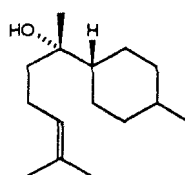
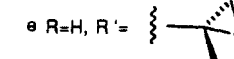
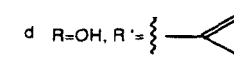
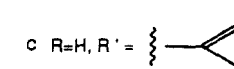
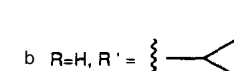
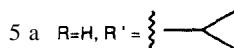
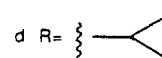
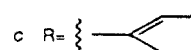
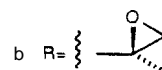
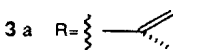
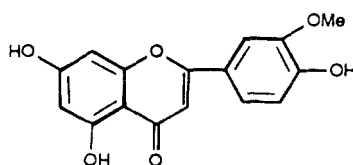
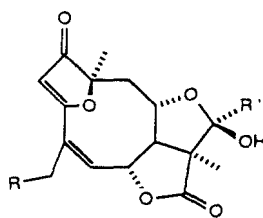


2 a $R=MeAcr$



f $R = T|g|$

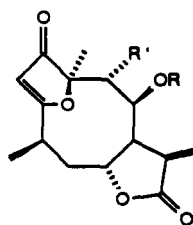
g R=Ang



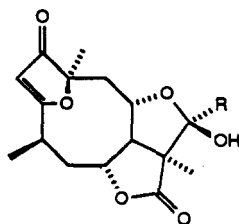
9 a R=MeAcr

b $R = T|g|$

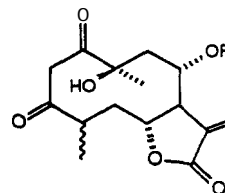
c $R=Ang$



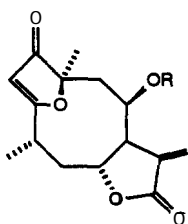
10a R=MeAcr, R'=H
 b R=Tigl, R'=H
 c R=Ang, R'=H
 d R=MeAcr, R'=OAc



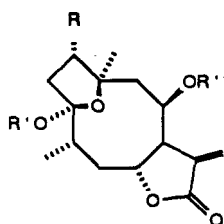
11a R=
 b R=
 c R=



12



13a R=MeAcr
 b R=Ang



14a R, R'=OH, R''=iBu
 b R, R'=H, R''=iBu
 c R=H, R'=Et, R''=iBu
 d R, R'=OH, R''=i-Val

in Table 1; comparison with the spectra of our four compounds from *E. seidelii* and *E. goyazensis* reveals subtle differences in the chemical shifts of H-5 $\alpha\beta$ and H-15 and in the coupling constants involving H-5a and H-5 β (which are essentially inverted) and the presence (in the previously reported lactones) or absence in the new lactones of allylic coupling between H-2 and H-4. Hence the two series differ in stereochemistry at C-4. While models of both **2a–d** and **9a–e** can be twisted to accommodate an H-2, H-4 dihedral angle requisite for allylic coupling between these two protons [33],* the observed strong enhancement of the H-2 signal (12.5%) on irradiation of the frequency of H-4 in our compounds seemed to argue in favour of formulas **2a–d**.

In a quite similar way it was shown that our lactones **3a, b** whose ^1H and ^{13}C NMR spectra are listed in Tables 3 and 4 were 4,5-dihydro derivatives of eremantholide C (**5c**) and its so far unknown 18,19-epoxide **5e**. Three such lactones originally formulated as **3a, 3c** and **3d** [7] and

later [32] revised to **11a–c** have been isolated from *E. bicolor*. For comparison the ^1H NMR spectrum of one of these, the presumed **11a**, is also listed in Table 3 and shows that the same spectral differences noted above between the 4,5-dihydro lactones from *E. seidelii* and *E. goyazensis*, on the one hand, and from *E. bicolor* on the other also exist for the 4,5-dihydroeremantholides. Consequently if the lactones from *E. bicolor* are indeed **11a–c**, those from *E. seidelii* and *E. goyazensis* should be **3a, b**.

To resolve once and for all the uncertainties which have plagued the assignment of C-4 stereochemistry in this series of compounds an X-ray analysis of **3b** was undertaken. Crystal data are given in the Experimental section. Figure 1 is a stereoscopic view of the molecule which shows that the C-4 methyl group of **3b**, and hence that of the other compounds from *E. seidelii* and *E. goyazensis*, is α orientated, i.e. that the C-4 stereochemistry is **4S** since the absolute configuration corresponds to the relative configuration shown in the formulas. Conversely in the compounds from *E. bicolor* and *P. ericoides* the C-4 methyl group is β orientated and C-4 is **4R**. Tables 5 and 6 list bond lengths and bond angles (the lists of final atomic and final anisotropic thermal parameters are deposited at the Cambridge Crystallographic Data Center).

In the crystal of **3b** the conformation of the I-membered ring embodying the 3,10-oxygen bridge is not

*A dihedral angle close to 90° required for maximum allylic coupling [33] is more easily achieved in models with H-4 α than in models with H-4 β , although the presence of allylic coupling was originally invoked [7] as an argument in favour of formulas **2a, f** for compounds now known to possess structures **9a, b**.

Table 1. ^1H NMR spectra of compounds

H	2a (CDCl_3)	2a (C_6D_6)*	2b (CDCl_3)	2b (C_6D_6)*	2c † (CDCl_3)	2c (C_6D_6)*	2d (CDCl_3)§	2d (C_6D_6)*
2	5.68 s	5.09 s	5.68 s	5.30 s	5.70 s	5.91 s	5.68 s	5.68 s
4	3.14 ddq (10, 6.5, 6.5)	2.01 ddq (11, 7, 7)	3.18 ddq (10.5, 7, 7)	2.35 ddq	3.29 ddq (11, 7, 7)	3.41 (obsc.)	3.25 ddq (11, 7, 7)	3.06 ddq (10.5, 7.5, 7)
5 α	1.94 ddd (13.5, 11, 10.5)	1.27 ddd	1.91 ddd (14, 11, 10.5)	1.31 ddd	1.89 ddd (14, 11, 10.5)	1.36 ddd	1.88 ddd (13, 11, 10.5)	1.35 ddd
5 β	2.45 dd (13.5, 7)	1.86 dd	2.43 dd (14, 7)	1.96 dd	2.46 dd (13.5, 7)	2.27 dd	2.45 br dd (13.5, 7)	2.16 hrdd (13.5, 7.5)
6	4.52 dd (11, 5)	4.37 dd	4.47 dd (10.5, 4.5)	4.35 dd	4.50 dd (10.5, 4.5)	4.60 dd	4.50 dd (11, 5)	4.53 dd
7	3.37 m	2.89 m	3.40 dddd (5, 3, 2.5, 2.5)	2.92 m	3.43 m	2.94 m	3.40 m	2.95 m
8	4.39 ddd (12, 3, 1.5)	4.46 ddd	4.33 ddd (11.5, 2.5, 2.5)	4.40 ddd	4.39 ddd (12, 3, 2)	4.49 ddd	4.33 ddd (12, 3, 1.5)	4.41 ddd
9 α	2.47 dd (13.5, 12)	1.86 dd	2.44 dd (14, 11.5)	1.84 dd	2.43 dd (13, 12.5)	1.80 dd	2.42 dd (13, 11.5)	1.88 dd
9 β	2.32 dd (14, 2)	2.16 dd	2.29 dd (14, 2)	2.14 dd	2.30 dd (13.5, 2)	2.10 dd	2.21 dd (13, 1.5)	2.02 dd
13a	6.21 d (3)	6.27 d	6.29 d (3)	6.23 d	6.31 d (3)	6.24 d (3)	6.29 d (3)	6.25 d
13b	5.45 d (2.5)	5.05 d	5.47 d (2.5)	5.04 d	5.49 d (2.5)	4.97 d	5.48 d (2.5)	5.05 d
14†	1.50 s	1.14s	1.45 s	1.16 s	1.48 s	1.29 s	1.47 s	1.23 s
15†	1.34 d (6.5)	0.71 d (7)	1.32 d (7)	1.16 s	1.32 d (7)	0.87 d	1.32 d (7)	0.84 d
3'a	6.00 br d (1.5)	5.87 br d	2.39 d (6)	2.41 d	3.62 d (11.5)	3.46 d	4.22 d (11.5)	4.28 d
3'b	5.52 dq (1.5)	5.07 (obsc.)	2.69 d (6)	2.04 d	3.50 d (12)	3.41 d	3.97 (11.5)	4.13 d
4†	1.83 br d (1.5)	1.63 br d	1.48 s	1.13 s	1.34 s	1.10s	1.27 s	1.10 s
OH	—	—	—	—	—	—	4.20 br	4.27 br
A c t	—	—	—	—	—	—	2.07 s	1.97 s

*All signals broadened.

†Intensity three protons.

‡No change on addition of TAI except for H-3'a 4.11 d, H-3'b 3.76 d, H-4' 1.61 s NH 8.60 s, 10.3 br

§No change on addition of TAI except for H-3'a 4.50 d, H-3'b 4.29 d, H-4' 1.57 s, NH 8.5s.

significantly different from that found for other 3,10-epoxy-4,5-dihydroheliangolides [34–37]. If the conformation of **3b** and its congeners in solution approximates the conformation in the solid state, the coupling constants, the pronounced NOE between H-2 and H-4 and the absence of allylic coupling between H-2 and H-4 are all accounted for. It is interesting that biosynthesis of these substances—whether it involves biocyclization of precursors such as **12** or enzymatic reduction of precursors of type 4 or 5 seems to produce different C-4 stereochemistry in *E.seidelii* and *goyazensis*, on the one hand, and *E.bicolor* and *P.ericoides* on the other, although the possibility of epimerization at C-4, the γ -position of an α -unsaturated ketone, possibly via a hemiketal-ketal equilibrium such as **2** \rightleftharpoons **12** \rightleftharpoons **9** during the extraction and isolation process cannot be totally dismissed.

In this connection it seems appropriate to comment on the confusion in the literature concerning the C-4 stereo-

chemistry of zexbrevin (**10a**) and its ester analogues **10b–d**.* Zexbrevin was originally isolated from *Zexmenia brevifolia* [40] and was subsequently also found in *Calea zacatechichi* together with its tiglate analogue **10b** [41] and in *Viguiera gregii* [42]. Its initially proposed stereochemistry **2a** [40] was revised to **13a** as a result of correlations involving the conversion of tagitinin A (**14a**) to tetrahydrozexbrevin and to tirotundin (tagitinin D, **14b**) [43, 44]. The stereochemistry assigned to tirotundin was based on an X-ray analysis of its ethyl ether **14c** [34] and formed the cornerstone of this complex set of interrelationships.

* Because of the confusion, Gao, Wang and Mabry [38] failed to recognize that their ladibranolide **10c** from *Viguiera latibracteata* was in fact identical with trichomoriolide from *Trichogoniopsis morii* [39] which was originally assigned structure **13c** in accordance with the then accepted formula for zexbrevin.

2a-2e and **9a-9c** (270 MHz)

2e (CDCl ₃)	2e (C ₆ D ₆)*	9a,b (CDCl ₃)	9c ** (CDCl ₃)	10a ** (CDCl ₃)	16b (CDCl ₃)	16b (C ₆ D ₆ +3 drops DMSO- <i>d</i> ₆)
5.70 s	5.00 s	5.70 d (1.5)	5.69 d	5.69 br s	5.56 br	2.88 d (18.5) 2.63 d (18.5) 2.80 d 2.53 d
3.17 ddq (11, 7, 7)	2.03 ddq	3.06 brdd (7, 7, 1.5)		3.05 br dq (7, 7)	3.05 br dq (1, 7, 7)	2.04 ddq (7) 1.52 dd
1.91 ddd (13, 11, 10.5)	1.26 ddd	2.11 brd (14, 1.5)		2.11 brd (14)	2.06 br d (15)	1.95 br d (14) 1.83 brd
2.43 dd (13.5, 7)	1.81 dd	2.49 ddd (14, 7, 11)		2.50 ddd (14, 7, 11)	2.61 ddd (15, 7, 8.5)	2.23 ddd (14, 11, 9) 2.45 ddd
4.42 dd (11, 5)	4.20 dd	4.34 ddd (11, 5, 1.5)	4.33 ddd	4.36 ddd (11, 5?)	4.48 dd (8.5, 5)	4.54 ddd (11, 6 ~ 1.5) 4.41 ddd
3.46 m	2.95 m	3.36 ddd (5.4, 3.5, 3)		3.35 m (5, 4, 3.3, 3)	3.30 br ddd (5, 3, 2.7)	4.09 dddd (6, 3, 3, 3) 4.15 ddd
4.40 ddd (12, 3, 1.5)	4.42 ddd	4.50 ddd (10, 4, 2)	4.51 brdd	4.53 ddd (10, 4, 2)	5.17 dd (5, 2.5)	5.61 ddd (11.5, 4.5, 3.3) 5.64 ddd
2.39 dd (13, 5, 12)	1.83 dd	2.35 dd (13.5, 10)	2.34 dd	2.34 dd (13, 10)	2.71 dd (16, 5)	1.70 dd (14, 11.5) 1.55 dd
2.24 dd (13.5, 2)	2.09 dd	2.48 dd (13.5, 2)		2.46 dd (13, 2)	2.25 dd (16, 2.5)	1.97 dd (14, 4.5) 1.96 dd
6.34 d (3)	6.24 d	6.19 d (3.5)	6.16 d	6.19 d (3.3)	6.36 d (3)	6.30 d (3) 6.22 d
5.52 d (2.5)	4.99 d	5.46 d (3)	5.43 d	5.42 d (3)	5.70 d (2.7)	5.58 d (3) 5.25 d
1.48 s	1.15 s	1.49 s	1.48 s	1.48 s	1.42 s	1.78 s 1.41 s
1.33 d (7)	0.68 d	1.42 d (7)		1.43 d (7)	1.39 d (7)	1.56 d (7) 0.96 d
3.99 d (11.5)	3.99 d				5.99 br d (1.5)	1.05 d† (7) 0.98 d† (7)
3.65 d (11.5)	3.44 d				5.60 dq (1.5, 1.5)	
1.52 s	1.40 s				1.87 brd (1.5)	1.07 d (3) 3.79 s 2.43 sept (7)-H-2 0.97 d (7) 3.40 s 2.26 sept (7)-H-2
2.05 s	1.63 s					

‖ Taken from ref. [7].

¶ Taken from ref. [8].

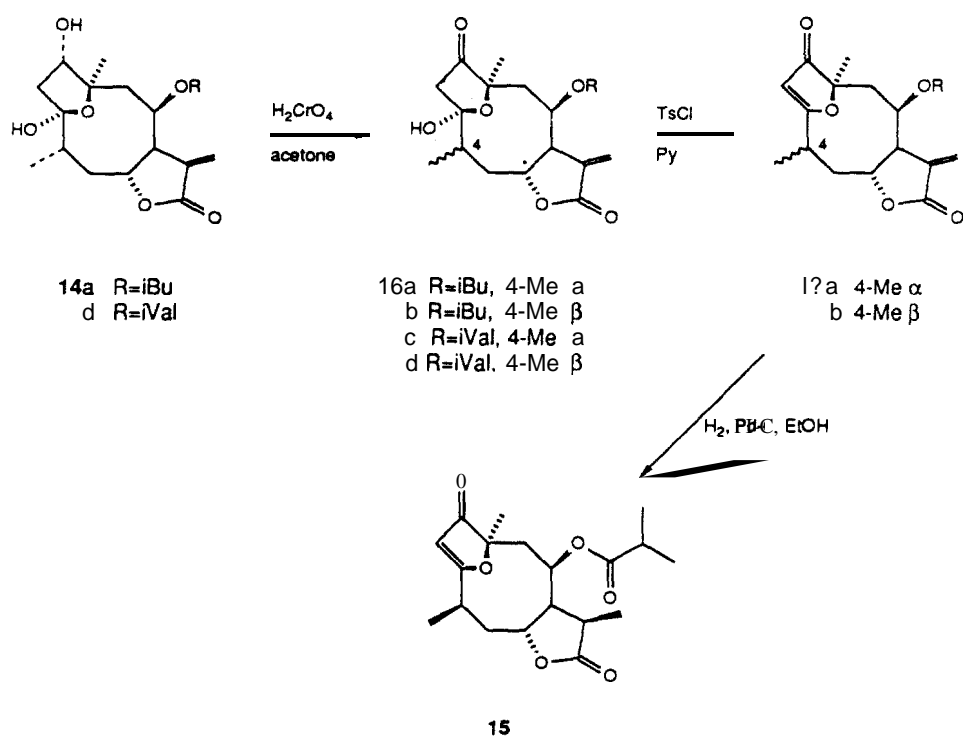
**At 200 MHz, taken from ref. [42].

Somewhat later isolation from a *Calea* species and subsequent X-ray analysis of a **lactone 10d** which had 'NMR properties reminiscent of a **9 α -acetoxyzexbrevin**' led Fischer and coworkers [35, 45] to reformulate **zexbrevin** as **10a**,† while in essentially contemporaneous articles on the photochemistry of tetrahydrozexbrevin [46] based largely on X-ray analyses of tetrahydrozex-

brevin [37] and phototetrahydrozexbrevin A [47] formulas of these compounds were consistently represented as containing an α -orientated C-4 methyl group.‡ Only careful study of Fig. 1 and 2 of ref. [37], the latter representing a stereoscopic view of the conformation and packing of tetrahydrozexbrevin, and Fig. 1 of ref. [47] reveals that the C-4 methyl groups of tetrahydrozexbrevin and phototetrahydrozexbrevin A are in fact β -orientated and that tetrahydrozexbrevin is actually represented by formula 15 (Scheme 1) in accordance with the proposal of the Fischer group. Indeed, the ¹H NMR spectrum of zexbrevin (Table 1) exhibits the features now known to be diagnostic for B-orientation of the C-4

†As is evident from the previous discussion NMR evidence was, at the time, in the absence of information about spectral properties of C-4 diastereomers, not sufficient to distinguish between α - and β -orientation of the C-4 methyl group. In retrospect the presence of allylic coupling between H-2 and H-4 evidenced by broadening of the H-2 signal which is exhibited by **zexbrevin** as well as by **9a-c**, **10b-d** and **11a-c**, but not by **2a-e** and **3a, b** can be seen as being diagnostic of β -orientation of the C-4 methyl group in zexbrevin.

‡In fact ref. [37] is misleadingly entitled '(4 β H,6 β H,11 α H)-3 β ,10 β -epoxy-8 β -isobutyryloxy-1-oxogermacr-2-en-6,12-olide (tetrahydrozexbrevin), C₁₉H₂₆O₆ Sesquiterpene Lactone'.



Scheme 1.

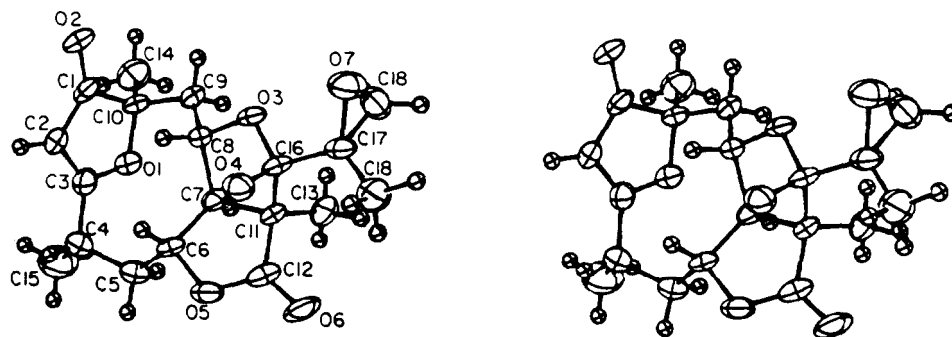
Table 2. ¹³C NMR spectra of compounds **2a–e** (CDCl₃, 67.89 MHz)*

C	2a	2b	2c	2d	2e	10a †
1	204.95 <i>s</i>	204.68 <i>s</i>	205.50 <i>s</i>	204.57 <i>s</i>	204.97 <i>s</i>	205.18 <i>s</i>
2	105.52 <i>d</i>	105.61 <i>d</i>	106.37 <i>d</i>	105.74 <i>d</i>	105.78 <i>d</i>	102.05 <i>s</i>
3	192.57 <i>s</i>	192.44 <i>s</i>	192.43 <i>s</i>	192.32 <i>s</i>	192.10 <i>s</i>	192.13 <i>s</i>
4	33.56 <i>d</i>	33.55 <i>d</i> †	33.00 <i>d</i>	33.29 <i>d</i>	33.49 <i>d</i> †	31.37 <i>d</i>
5	42.35 <i>t</i>	42.34 <i>t</i> t	42.22 <i>t</i>	42.30 <i>t</i>	42.30 <i>t</i>	40.96 <i>t</i>
6	82.14 <i>t</i>	82.05 <i>d</i> t	81.99 <i>d</i>	82.07 <i>d</i>	82.13 <i>d</i>	74.41 <i>d</i>
7	54.36 <i>d</i>	54.16 <i>d</i> †	53.75 <i>d</i>	53.94 <i>d</i>	53.78 <i>d</i>	51.61 <i>d</i>
8	71.93 <i>d</i>	75.21 <i>d</i> †	73.52 <i>d</i>	73.31 <i>d</i>	12.74 <i>d</i>	74.41 <i>d</i>
9	45.23 <i>t</i>	45.11 <i>t</i> †	45.21 <i>t</i>	45.22 <i>t</i>	45.17 <i>t</i> †	43.10 <i>t</i>
10	89.84 <i>s</i>	89.78 <i>s</i>	89.59 <i>s</i>	89.67 <i>s</i>	89.61 <i>s</i>	88.29 <i>s</i>
11	133.93 <i>s</i>	133.59 <i>s</i>	133.15 <i>s</i>	132.92 <i>s</i>	132.77 <i>s</i>	139.53 <i>s</i>
12	168.72 <i>s</i>	168.33 <i>s</i>	168.77 <i>s</i>	168.42 <i>s</i>	168.65 <i>s</i>	168.54 <i>s</i>
13	124.07 <i>t</i> †	124.24 <i>t</i> †	125.60 <i>t</i>	124.92 <i>t</i>	125.12 <i>t</i>	123.05 <i>t</i>
14	21.07 <i>q</i> †	21.06 <i>q</i>	21.21 <i>q</i>	21.09 <i>q</i>	20.91 <i>q</i>	22.74 <i>q</i>
15	18.46 <i>q</i> †	18.44 <i>q</i> †	18.16 <i>q</i>	18.36 <i>q</i>	18.56 <i>q</i>	16.04 <i>q</i>
1'	166.74 <i>s</i>	170.27 <i>s</i>	173.50 <i>s</i>	174.05 <i>s</i>	174.65 <i>s</i>	166.12 <i>s</i>
2	135.61 <i>s</i>	53.52 <i>s</i>	74.30 <i>s</i>	73.40 <i>s</i>	78.77 <i>s</i>	135.82 <i>s</i>
3'	126.10 <i>t</i> †	52.51 <i>t</i>	50.88 <i>t</i>	68.85 <i>t</i>	46.74 <i>t</i> †	126.70 <i>t</i>
4'	17.87 <i>q</i>	17.23 <i>q</i>	22.82 <i>q</i>	20.49 <i>q</i>	21.65 <i>q</i> †	18.05 <i>q</i>
AC				170.62 <i>s</i>	171.06 <i>s</i>	
				22.04 <i>q</i>	20.65 <i>q</i> †	

*Multiplicities established by DEPT pulse sequence.

† Assignment by single frequency heteronuclear decoupling.

‡ Taken from ref. [41].

Fig. 1. Stereoscopic view of compound **36**.Table 3. ^1H NMR spectra of compounds **3a**, **b** and **11a** (CDCl_3 , 270 MHz)

H	3a	3a*	3b	3b*	11af
2	5.57 s	5.29 s	5.59 s	5.87 s	5.61 brs (1.5)
4	3.12 ddq (11, 7, 7)	2.50 ddq (obsc)	3.13 ddq (11, 7, 7)	3.04 ddq	3.00 br dq (7, 7, 1.5, 1.5)
5 α	1.86 ddd (13, 11, 10.5)	1.55 ddd	1.86 ddd (13, 11, 10.5)	1.55 ddd	2.15 br d (14, 1.5)
5 β	2.47 br dd (13, 7)	2.14 br dd	2.49 ddd (13, 7)	2.40 dd	2.44 dd (14, 11, 7)
6	4.27 dd (11, 7)	4.18 dd	4.28 dd (13.5, 7)	4.41 dd	4.24 dd (11, 6.5)
7	2.53 dd (7, 4.5)	2.26 dd	2.57 dd (7, 4.5)	2.26 dd (7, 5)	2.57 dd (6.5, 5)
8	3.99 ddd (12, 4.5, 2)	4.01 ddd	3.98 ddd (12, 5, 2)	4.05 ddd	3.98 ddd (12, 5, 2)
9a	1.96 dd (13, 12)	1.70 dd	1.90 dd (13, 12)	1.68 dd	2.46 dd (13.5, 2)
9 β	2.44 dd (13, 2)	2.50 dd	2.38 dd (13, 2)	2.61 dd	2.01 dd (13.5, 12)
13 \dagger	1.20 s	1.13 s	1.37 s	1.08 s	1.22 s
14 \dagger	1.43 s	1.15 s	1.48 s	1.39 s	1.45 s
15 \dagger	1.31 d (7)	0.86 d	1.31 d (7)	1.08 s	1.38 d (7)
18a	5.30 br s	3.43 br s	3.12 d (6)	2.98 d (6)	5.23 br s
18b	5.05 t (1)	5.05 br s	2.71 d (6)	2.38 d (6)	5.07 br s
19 \dagger	1.90 br s	2.89 br s	1.57 s	1.78 s	1.90 br s
OH	2.75 s	4.05 br	3.31 s		1.71 s

*In C_6D_6 solution. All signals broadened. \dagger Intensity 3 protons. \ddagger Taken from ref. [7].

methyl group (allylic coupling between H-2 and H-4, relative shifts of H-5 α and H-5 β and their coupling constants).

Stereochemistry **15** for tetrahydrozexbrevin implies that during stage 1 or 2 of the three stage conversion of tagitinin A to tetrahydrozexbrevin (Scheme 1) [43] the asymmetric centre at C-4 was inverted or that tagitinin A also possesses a γ -orientated C-4 methyl group. In the latter case C-4 epimerization, presumably by equilibration through a hemiacetal- γ -ketol equilibrium, must have

taken place during the conversion [43] of tagitinin A to tirotundin **14b** and its ethyl ether **14c**.

Indirect but persuasive evidence has been adduced recently [48, 49] in favour of our original formulation of tagitinin A as **14a** so that epimerization during the conversion of **14a** to **15a** (Scheme 1) seemed *a priori* more likely to account for the C-4 stereochemistry of zexbrevin than a change in the C-4 configuration of tagitinin A. Superposition of the signals of H-2b, H-4 and H-5a, b in the ^1H NMR spectrum of tagitinin A interfered with accu-

Table 4. ^{13}C NMR spectra of compounds **3a**, **b** (CDCl_3 , 67.89 MHz)*

C	3a	3b
1	205.81 <i>s</i>	205.75 <i>s</i>
2	105.09 <i>d</i>	105.10 <i>d</i>
3	192.58 <i>s</i>	192.47 <i>s</i>
4	33.41 <i>d</i>	33.18 <i>dt</i>
5	43.98 <i>t</i>	43.63 <i>t t</i>
6	81.54 <i>d</i>	81.55 <i>dt</i>
7	66.40 <i>d</i>	67.11 <i>d†</i>
8	76.90 <i>d</i>	76.49 <i>d t</i>
9	42.26 <i>t</i>	42.18 <i>t†</i>
10	90.38 <i>s</i>	90.25 <i>s</i>
11	60.21 <i>s</i>	60.04 <i>s†</i>
12	175.81 <i>s</i>	174.97 <i>s</i>
13	22.59 <i>q</i>	21.32 <i>qt</i>
14	21.04 <i>q</i>	20.94 <i>q†</i>
15	18.93 <i>q†</i>	18.48 <i>q</i>
16	106.43 <i>s</i>	104.94 <i>s</i>
17	142.25 <i>s</i>	58.39 <i>s†</i>
18	115.78 <i>t</i>	53.64 <i>t†</i>
19	18.46 <i>q†</i>	17.33 <i>qt</i>

*Multiplicities established by DEPT pulse sequence.

†Assignment by single frequency heteronuclear decoupling.

‡Assignments interchangeable within column.

Table 5. ^{13}C NMR spectra of compounds **4b**, **c** (CDCl_3 , 67.89 MHz)*

C	4b	4c
1	204.65 <i>s</i>	204.71 <i>s</i>
2	104.65 <i>d</i>	104.69 <i>dt</i>
3	186.78 <i>s</i>	186.83 <i>s</i>
4	130.42 <i>s†</i>	130.44 <i>s†</i>
5	135.09 <i>d</i>	135.12 <i>d</i>
6	81.46 <i>d</i>	81.61 <i>dt</i>
7	51.19 <i>d</i>	51.23 <i>dt</i>
8	73.40 <i>d</i>	73.00 <i>d†</i>
9	43.81 <i>t</i>	44.04 <i>t†</i>
10	89.55 <i>s</i>	89.64 <i>s</i>
11	133.62 <i>s†</i>	133.84 <i>s†</i>
12	168.66 <i>s</i>	168.02 <i>s</i>
13	124.30 <i>t</i>	124.10 <i>t</i>
14	20.62 <i>q</i>	20.65 <i>q t</i>
15	20.25 <i>q</i>	20.22 <i>q†</i>
1'	166.69 <i>s</i>	167.00 <i>s</i>
2	135.58 <i>s</i>	126.51 <i>s</i>
3'	126.28 <i>t</i>	140.54 <i>d</i>
4	17.92 <i>q</i>	19.93 <i>q</i>
5'		15.64 <i>q</i>

*Multiplicities established by DEPT pulse sequence.

†Assignment by single frequency heteronuclear decoupling.

‡Assignments interchangeable within column

rate determination of the relevant coupling constants; however reexamination of a small sample of dehydrotagitin A (**16a** or **16b**) remaining from our earlier work [43] on a modern NMR spectrometer resulted in the frequencies and coupling constants listed in Table 1. It is seen that the coupling constants involving H-4, H-5 and H-6 essentially duplicate those found in zexbrevin (**10a**) rather than those found in the C-4 epimers **2a-e**. Consequently we believe that dehydrotagitin A is best described by formula **16b** and that epimerization of **16a** to the more stable [49]4 β -methyl isomer **16b** is at present the most plausible explanation for the correlations involving zexbrevin, tagitin A and tirotundin. The analogous conversion [SO] of viguilenin (**14d**) to dehydroviguilenin (**16d**) presumably follows the same course since the ^1H and ^{13}C NMR spectra of **14a** and **14d** on the one hand and **16b** and **16d** are superimposable except for the signals of the differing side chains.

EXPERIMENTAL

Plant material. Aerial parts of *Eremanthus seidelii* MacLeish and Schumacher were collected by W. V. in October 1985 in Furnas near Capitolio, Minas Gerais State. Aerial parts of *Eremanthus goyazensis* Sch.-Bip. were collected by W. V. in Furnas in May 1985. *Vanillosmopsis erythropappa* Sch.-Bip. wood was collected at the same location: vouchers are kept in the FCF de Ribeirão Preto.

Extraction and isolation of constituents. Aerial parts, mainly leaves, of *E. seidelii* (2.95 kg) were pulverized and extracted with

EtOAc to give 122 g of crude extract which was dissolved in $\text{MeOH-H}_2\text{O}$ (19: 1). The soln was extracted with hexane: the $\text{MeOH-H}_2\text{O}$ layer was concd at red. pres. and the residue extracted with EtOAc. Evapn of the extract at red. pres. gave 68 g of residue which was chromatographed over 400g silica gel, 200 ml fractions being eluted as follows: F1-10 (hexane), 11-48 (hexane-EtOAc 99: 1), 49-60 (hexane-EtOAc 49: 1), 61-72 (hexane-EtOAc 24: 1), 73-87 (hexane-EtOAc 47: 3), 88-95 (hexane-EtOAc 93: 7), 96-120 (hexane-EtOAc 9: 1), 121-137 (hexane-EtOAc 17: 3), 138-159 (hexane-EtOAc 4: 1), 160-190 (hexane-EtOAc 3: 2), 224-235 (hexane-EtOAc 1: 1), 236-250 (hexane-EtOAc 2: 3), 256-265 (hexane-EtOAc 3: 7), 266-280 (EtOAc), 281-300 (EtOH).

Recrystallization of fr. 95-98 from hexane-EtOAc (4: 1) gave 640 mg of **2a**, mp 174-176°. Recrystallization of fr 104-111 from hexane-EtOAc (7: 3) gave 1.8 g of (**3a**) mp 228-232°. Frs 128-144 on recrystallization from hexane-EtOAc gave 1.0 g of **3b**, mp 225°. Recrystallization of frs 145-155 from EtOH gave 110 mg of isorhamnetin (**6**), identified (^1H NMR, MS) by comparison with an authentic sample. A 231 mg portion of frs 156161 (6.0 g) on purification by TLC (hexane-EtOAc 7: 3, twice) afforded 160 mg of **2d** and 60 mg of a 4: 1 mixture of **2c** and **2b**, which was separated by acetylation of the mixture and prep. TLC (*vide infra*).

Aerial parts of *E. goyazensis* (3.2 kg) were pulverized and extracted with EtOAc to give 200 g of crude extract which was dissolved in $\text{MeOH-H}_2\text{O}$ (195: 5). The soln was extracted with hexane; the $\text{MeOH-H}_2\text{O}$ layer was concd at red. pres. and the residue extracted with EtOAc. Evapn of the extract gave 80 g of residue which was chromatographed over 400 g silica gel, 400 ml

Table 6. Bond lengths (Å) in **3b** with standard deviations in parentheses.

Distances	Bond length (Å)
C1–C2	1.409 (9)
C1–C10	1.522 (9)
C2–C3	1.340 (9)
C3–O1	1.343 (9)
C3–C4	1.503 (9)
C4–C5	1.522 (10)
C4–C15	1.538 (7)
C5–C6	1.453 (7)
C6–O5	1.543 (7)
C6–C7	1.542 (7)
C7–C8	1.532 (7)
C7–C11	1.447 (7)
C8–O3	1.495 (7)
C8–C9	1.527 (7)
C9–C10	1.445 (7)
C10–O1	1.522 (7)
C10–C1	1.511 (7)
C11–C12	1.530 (8)
C11–C16	1.564 (7)
C12–C13	1.530 (7)
C12–O5	1.336 (7)
C12–O6	1.217 (7)
C16–O3	1.408 (7)
C16–O4	1.395 (7)
C16–C17	1.510 (7)
C17–O7	1.442 (7)
C17–C19	1.447 (9)
C17–C18	1.463 (8)
C18–O7	1.432 (8)

fractions being eluted as follows: Frs 1–7 (hexane), **8:13** (hexane–EtOAc 9: 1), 14–24 (hexane–EtOAc 19: 1), 25–47 (hexane–EtOAc 9: 1), 48–69 (hexane–EtOAc 17: 3), **70–86** (EtOAc 4: 1), 87–100 (hexane–EtOAc 3: 1), 101–112 (hexane–EtOAc 7: 3), 113–127 (hexane–EtOAc 3: 2), **128–140** (hexane–EtOAc 2: 3), 141–156 (hexane–EtOAc 1: 4), 157–168 (EtOAc), 169–173 (EtOAc–EtOH 1: 1), 174–181 (EtOH).

Frs 60–85 were combined and recrystallized (hexane–EtOAc 7: 3) to give 300 mg of eremantholide C (**5c**), mp 232° lit mp 229–230° [20], identical in all respects with an authentic sample. Frs **86–100** were combined and recrystallized (hexane–EtOAc 4: 1) to give 191 mg of **2a**. Frs 111 and 112 were combined and recrystallized from hexane–EtOAc (7: 3) to give 60 mg of **3b**, 143–144°. Frs 113–120 (combined wt **5 g**) were mixtures of goyazensolide (**4a**) [17,18] and **5d** [9]. Frs **121–132** were combined and washed with a little CHCl₃. The CHCl₃-insoluble material (1.4 g) was slightly impure **2c**. The CHCl₃ soluble material (2.0 g) was rechromatographed over silica gel, elution with hexane–EtOAc containing increasing amounts of EtOAc. This initially afforded 149 mg of a mixture containing **3b**, **2b** and **2c**; the more polar eluates were essentially pure **5d**.

Frs **128–140** (1.12 g) were combined and rechromatographed over 20 g of silica gel inactivated with 10% H₂O. The hexane–EtOAc (7: 3) eluate was purified to give in order of increasing polarity **3b**, **2c** and **2b**. Frs 141–156 (1.0 g) were also combined and rechromatographed over 20 g silica gel. Elution with hexane–EtOAc mixtures of increasing polarity afforded, in order, 25 mg of isorhamnetin and 220 mg of a mixture which was

Table 7. Bond angles (°) in **3b** with standard deviations in parentheses

Atoms	Angles (deg)
c3 O1 C10	108.4 (7)
C8 O3 C16	107.9 (7)
O2 C1 C10	121.6 (9)
O2 C1 c2	130.2 (9)
C10 C1 c2	108.1 (9)
c3 c2 C1	106.2 (9)
C6 C5 C12	112.2 (10)
O1 c3 c4	113.2 (9)
O1 c3 c2	115.0 (9)
c4 c3 c2	131.7 (9)
c3 c4 C5	111.5 (7)
c3 c4 C15	110.8 (7)
C5 c4 C15	111.1 (7)
C4 C5 C6	115.3 (7)
O5 C6 C5	107.9 (7)
O5 C6 C7	105.2 (7)
C5 C6 C7	115.9 (7)
C6 C7 C8	116.7 (7)
C6 C7 C11	105.4 (7)
C8 C7 C11	105.9 (7)
O3 C8 C7	104.6 (7)
O3 C8 C9	107.5 (7)
C7 C8 C9	116.9 (7)
C8 C9 C10	113.6 (7)
O1 C10 c9	106.8 (7)
O1 C10 C1	102.0 (7)
O1 C10 C14	110.1 (7)
c9 C10 C1	109.6 (7)
c9 C10 C14	114.5 (7)
C1 C10 C14	113.0 (7)
c7 C11 C12	103.3 (8)
c7 C11 C13	113.4 (7)
c7 C11 C16	100.8 (7)
C12 C11 C13	111.6 (7)
C12 C11 C16	113.1 (7)
O13 C11 C16	113.7 (7)
O5 C12 O6	121.4 (7)
O5 C12 C11	111.8 (7)
O6 C12 C11	126.7 (7)
O3 C16 O4	110.9 (7)
O3 C16 C11	103.2 (7)
O3 C16 C17	108.8 (7)
O4 C16 C11	105.3 (7)
O4 C16 C17	111.7 (7)
C11 C16 C17	116.6 (7)
C16 C17 O7	114.3 (7)
C16 C17 C19	116.0 (7)
C16 C17 C18	119.2 (7)
O7 C17 C19	114.0 (7)
O7 C17 C18	59.1 (8)
C19 C17 C18	120.8 (9)
C17 O7 C18	61.2 (8)
C17 C18 O7	59.7 (8)

separated by HPLC (Lichnisorb RP-18, 5 µm, solvent MeOH–H₂O 7: 3, flow rate 0.5 ml/min) into goyazensolide (**4a**) and its **11β** (H), **13-dihydro** derivative (**4d**). Frs 157–168 (4 g) were also combined. Rechromatography of the material over silica gel

(eluent hexane-EtOAc 2:3) failed to separate the constituents which were identified as **3a**, **4a** and **5d**.

Powdered wood of *Vanillosmopsis erythropappa* (9.8 kg) was extracted with hexane. The constituents of the hexane extract or wood oil have been reported [1, 4, 8, 29]. Further extraction of the wood with EtOAc and evapn of the extract afforded 240 g of material which was chromatographed over 400g silica gel, 200 ml fractions being collected as follows: frs 1-10 (hexane), 11-20 (hexane-EtOAc 9:1), 21-60 (hexane-EtOAc) 61-80 (hexane-EtOAc 4:1), 81-110 (hexane-EtOAc 7:3), 111-125 (hexane-EtOAc 3:2), 126-140 (hexane-EtOAc 1:1), 141-160 (hexane-EtOAc 2:3), 161-170 (hexane-EtOAc 1:4), 171-190 (EtOAc) and 191-200 (EtOH). Fr. 5 contained 135 mg of bisabolol. Recrystallization of frs 90-105 (3.0 g) from hexane-EtOAc (7:3) gave 1.4 g of lychnopholide (**4c**), mp 125-128°, lit mp 128° [12] (for correction of structure see [8]). Fr. 115 (0.8 g) after purification by prep. TLC afforded costunolide and a very small amount of **4b**. Combination of frs 120-130 (1.0 g) and chromatography over silica gel followed by prep. TLC (CH₂Cl₂-EtOAc 3:2) gave additional **4b**. The previously unreported ¹³C NMR spectra of **4b** and **4c** are listed in Table 5.

(4S,6R,7S,8S,10R)-1-Oxo-3,10-epoxy-8-methacryloxygermacra-2,11(13)-dien-6,12-olide (**2a**). Mp 174-176°; (hexane-EtOAc 4:1); IR ν_{\max} cm⁻¹: 1765, 1710, 1695, 1593; MS EI m/z (rel. int.): 346 [M]⁺ (36.5), 277 (11.0), 260 (10.5), 249 (5.0), 232 (25.8), 217 (6.4), 190 (7.7), 165 (5.2), 137 (6.6), 125 (67.5), 69 (100); MS CI m/z : 347 [M+1]⁺ (100). ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

(4S,6R,7S,8S,10R)-1-Oxo-3,10-epoxy-8-(2,3-epoxypropanoyloxy)-germacra-2,11(13)-diene-6,12-olide (**2b**). Mp 143-144° (dec) (hexane-EtOAc 7:3); IR ν_{\max} cm⁻¹: 1770, 1740, 1710, 1605; MS EI m/z (rel. int.): 362 [M]⁺ (11.2), 277 (1.2), 260 (14.9), 232 (28.7), 125 (100), 57 (27); MS CI m/z : 363 [M+1]⁺ (100), 261 (3.4); ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

(4S,6R,7S,8S,10R)-1-Oxo-3,10-epoxy-8-(2,3-dihydroxypropanoyloxy)-germacra-2,11(13)-dien-6,12-olide (**2c**). Mp 175-177° (EtOAc-hexane 7:3); IR ν_{\max} cm⁻¹: 3400, 1770, 1740, 1695, 1600; MS EI m/z (rel. int.): 380 [M]⁺ (0.06), 362 (1.8), 349 (6.5), 277 (1.6), 261 (10.6), 260 (17.3), 232 (32.1), 181 (13.6), 149 (14.5), 125 (100); MS CI m/z : 381 [M+1]⁺ (3.7), 363 (20.9), 261 (5.6). ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

Acetylation of 30 mg of the mixture of **2b** and **2c** from *E. seidelii* (Ac₂O-pyridine, overnight) and purification of the crude product by prep. TLC (CH₂Cl₂-Et₂O 17:3) gave 13 mg of **2e** and 5 mg of **2b**. Compound **2e** was a gum; IR ν_{\max} cm⁻¹: 3480, 1760, 1745, 1705, 1600; MS EI m/z (rel. int.): 362 [M-C₂H₄O₂]⁺ (0.5), 349 (2.9), 277 (3.4), 278 (10.3), 261 (6.4), 260 (16.5), 232 (63.6), 190 (20.4), 181 (10.2), 163 (44.5), 149 (15.8), 135 (62.9), 125 (100), 117 (3.6), 112 (3.6), 99 (5.7); MS CI m/z (rel. int.): 423 [M+1]⁺ (0.5), 407 (6.5), 363 (5.7), 347 (18.2), 263 (29.8), 181 (16.8), 101 (13.7), 85 (7.2). ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

(4S,6R,7S,8S,10R)-Oxo-3,10-epoxy-8-(3-acetoxy-2-hydroxypropanoyloxy)-germacra-2,11(13)-diene-6,12-olide (**2d**). Colourless gum; IR ν_{\max} cm⁻¹: 3440, 1765, 1740, 1700, 1600; MS EI m/z (rel. int.): 422 [M]⁺ (2.9), 362 (14.4), 349 (14.7), 332 (2.6), 306 (23.7), 278 (20.1), 277 (2.2), 262 (38.1), 261 (15.9), 260 (20.3), 232 (48), 190 (15.6), 181 (23.5), 165 (6.5), 143 (19.2), 138 (11.9), 125 (100), 117 (26.8), 112 (17.2), 99 (46.3); MS CI m/z (rel. int.): 423 [M+1]⁺ (100), 401 (5.7), 360 (20.1), 347 (7.6), 279 (20.9), 263 (47.1), 261 (12.4), 189 (5.4), 163 (26.6), 145 (10.7), 117 (12.6), 107 (14.1). ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

(4S,6R,7S,8S,10R,11S,16R)-4,5-Dihydroeremantholide **C** (**3a**). Mp 2288232° (hexane-EtOAc 7:3); IR ν_{\max} cm⁻¹: 3360, 1766, 1694, 1589, 1215; MS EI m/z (rel. int.): 348 [M]⁺ (6), 331 (61), 330 (S), 279 (2), 236 (5.3), 168 (15.4), 136 (28.8), 125 (100), 69 (93.7); MS

CI m/z (rel. int.): 349 [M+1]⁺ (46.6), 332 (18.9), 331 (100); CD curve (MeOH) [θ]₃₀₈ + 2240, [θ]₂₇₆ 0, [θ]₂₆₀ - 964, [θ]₂₃₄ 0, [θ]₂₀₅ 0 (last reading). ¹H and ¹³C NMR spectra are listed in Tables 3 and 4.

(4S,6R,7S,8S,10R,11S,16R,17R)-4,5-Dihydro-17,18-epoxyeremantholide **C** (**3b**). Mp 225° (hexane-EtOAc); IR ν_{\max} cm⁻¹: 3355, 1754, 1686, 1592, 1245; MS EI m/z (rel. int.): 364 [M]⁺ (4.1), 347 (3.2), 346 (0.9), 307 (4.5), 236 (5.5), 125 (100), 85 (3.9), 69 (30.9); MS CI m/z : 365 [M+1]⁺ (100), 347 (33.7); CD curve (MeOH) [θ]₃₁₀ + 2230, [θ]₂₇₆ 0, [θ]₂₅₈ - 784, [θ]₂₃₉ 0, [θ]₂₀₅ 0 (last reading). ¹H and ¹³C NMR spectra are listed in Tables 3 and 4.

X-Ray analysis of 3b. Single crystals of **3b**, prepared by slow evapn of a 1:1 soln of hexane-EtOAc, were monoclinic, space group *P*₂₁ with *a* = 7.110(3), *b* = 10.304(7), *c* = 12.984(8) Å, β 74.96(7)° and *d*_{calcd} = 1.32 g/cm³ for *Z* = 2 (C₁₉H₂₄O₇, *M*_r = 364.4). The intensity data were measured on a CAD4 Enraf Nonius diffractometer (Mo radiation, monochromated, θ -2 θ scans). The size of the crystal used for data collection was ca 0.3 × 0.3 × 0.3 mm³. No absorption correction was necessary (μ = 0.94). A total of 1864 reflections were measured for 2 θ ≤ 50° of which 1572 were considered to be observed. The structure was solved by direct methods using MULTAN 78 [51] and refined by full-matrix least squares methods. In the final refinement anisotropic thermal parameters were used for non-hydrogen atoms. Methyl hydrogen atoms were located from a difference Fourier map; the remaining hydrogen atom parameters were calculated assuming idealized geometry. Hydrogen atoms were included in the structure factor calculations, but their parameters were not refined. The final discrepancy indices were *R* = 6.9 and *R*_w = 7.4 for the 1522 observed reflections. The final difference Fourier map was essentially featureless with no peak greater than 0.3 e/Å³.

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REFERENCES

- Baker, P. M., Fortes, C. C., Fortes, E. G., Gazzinelli, G., Gilbert, B., Calligari Lopes, J. N., Pellegrino, J., Tomassini, T. C. B. and Vichniewski, W. (1972) *J. Pharm. Pharmacol.* **24**, 853.
- Tomassini, T. C. B. and Gilbert, B. (1972) *Phytochemistry* **11**, 1177.
- Vichniewski, W. and Gilbert, B. (1972) *Phytochemistry* **11**, 2563.
- Garcia, M., DaSilva, A. J. R., Baker, P. M., Gilbert, B. and Rabi, J. A. (1976) *Phytochemistry* **15**, 331.
- Corbella, A., Gariboldi, P., Jommi, G. and Ferrari, G. (1974) *Phytochemistry* **13**, 459.
- Herz, W., Kumar, N., Vichniewski and W. Blount, J. F. (1980) *J. Org. Chem.* **45**, 2503.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1980) *Phytochemistry* **19**, 2663.
- Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1981) *Phytochemistry* **20**, 731.
- Bohlmann, F., Wallmeyer, M., King, R. M. and Robinson, H. (1982) *Phytochemistry* **21**, 1439.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1980) *Phytochemistry* **19**, 2669.
- Bohlmann, F., Müller, L., King, R. M. and Robinson, H. (1981) *Phytochemistry* **20**, 1149.
- Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1980) *Phytochemistry* **19**, 2381.
- Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1982) *Phytochemistry* **21**, 685.

14. Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1982) *Phytochemistry* 21, 1087.
15. Bohlmann, F., Brindöpke, G. and Rastogi, R. (1978) *Phytochemistry* 17, 415.
16. Bohlmann, F., Singh, P., Borthakur, N. and Jakupovic, J. (1981) *Phytochemistry* 20, 2379.
17. Vichneswski, W., Sarti, S. J., Gilbert, B. and Herz, W. (1976) *Phytochemistry* 15, 191.
18. Herz, W., and Goedken, V. L. (1982) *J. Org. Chem.* **47**, 2798.
19. Raffauf, R. F., Huang, R. K., LeQuesne, P. W., Levery, S. B. and Brennan, T. F. (1975) *J. Am. Chem. Soc.* **97**, 6884.
20. LeQuesne, P. W., Levery, S. D., Menachery, M.D., Brennan, T. F. and Raffauf, R. F. (1978) *J. Chem. Soc. Perkin I*, 1572.
21. Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1981) *Phytochemistry* 20, 1519.
22. Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1981) *Phytochemistry* 20, 739.
23. Bohlmann, F., Gupta, R. K., Jakupovic, J., Robinson, H. and King, R. M. (1981) *Phytochemistry* 20, 1609.
24. LeQuesne, P. W., Menachery, M. D., Pastore, M. P., Kelley, C. J., Brennan, T. F., Onan, K. D., Raffauf, R. F. and Weeks, C. M. (1982) *J. Org. Chem.* 47, 1519.
25. Bohlmann, F., Zdero, C., Robinson, H. and King, R. M. (1982) *Phytochemistry* 21, 1093.
26. Bohlmann, F., Singh, P., Zdero, C., Ruhe, A., King, R. M. and Robinson, H. (1982) *Phytochemistry* 21, 1669.
27. Barros, D. A. D., Calligari Lopes, J. L., Vichnewski, W., Calligari Lopes, J. N., Kulanthaivel, P. and Herz, W. (1985) *Planta Med.* 35.
28. MacLeish, N. F. F. and Schumacher, H. (1984) *System. Botany* 9, 84.
29. Lima, P. D. D. B., Garcia, M. and Rabi, J. A. (1985) *J. Nat. Prod.* 48, 986.
30. Vichnewski, W., Calligari Lopes, J. N., Santos Filho, D. D. and Herz, W. (1976) *Phytochemistry* **15**, 1775.
31. Gottlieb, O. and Magellaes, M. (1958) *Perfum. Essent. Oil Rec.* 49, 711.
32. Banerjee, S., Schmeda-Hirschmann, G., Castro, V., Schuster, A., Jakupovic, J. and Bohlmann, F. (1986) *Planta Med.* 29.
33. Barfield, M., Spear, R. J. and Sternhell, S. (1976) *Chem. Rev.* 76, 595.
34. Herz, W. and Blount, J. F. (1978) *J. Org. Chem.* 43, 1268.
35. Fronczek, F. R., Lee, I.-Y. and Fischer, N. H. (1983) *J. Nat. Prod.* 46, 104.
36. Fischer, N. H., Lee, L.-Y., Fronczek, F. R., Chiari, G. and Urbatsch, L. E. (1984) *J. Nat. Prod.* 47, 419.
37. Soriano-Garcia, M. and Toscano, R. A. (1984) *Acta Cryst. C* 40, 1425.
38. Gao, F., Wang, H. and Mabry, T. J. (1987) *Phytochemistry* 26, 779.
39. Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1982) *Phytochemistry* 21, 2035.
40. Romo de Vivar, A., Guerrero, C., Diaz, E. and Ortega, A. (1970) *Tetrahedron* **26**, 1675.
41. Herz, W. and Kumar, N. (1980) *Phytochemistry* 19, 593.
42. Liu, Y.-L., Gershenzon, J. and Mabry, T. J. (1984) *Phytochemistry* 23, 1967.
43. Baruah, N. C., Sharma, R. P., Madhusudanan, K. P., Thyagarajan, G., Herz, W. and Murari, R. (1979) *J. Org. Chem.* 44, 1831.
44. Chowdhury, P. C., Sharma, R. P., Thyagarajan, G., Herz, W. and Govindan, S. V. (1980) *J. Org. Chem.* **45**, 4993.
45. Lee, I.-Y., Fronczek, F. R., Malcolm, A., Fischer, N. H. and Urbatsch, L. E. (1982) *J. Nat. Prod.* 45, 310.
46. Rodriguez-Hahn, L., Jimenez, R., Sancedo, R., Soriano-Garcia, M., Toscano, R. A. and Diaz, E. (1983) *Tetrahedron* 39, 3909.
47. Soriano-Garcia, M., Toscano, R. A., Diaz, E. and Rodriguez-Hahn, L. (1985) *Rev. Latinoam. Quim.* 16, 112.
48. Sarma, J. C., Barua, N. C., Sharma, R. P. and Barua, J. N. (1983) *Tetrahedron* 39, 2843.
49. Sarma, J. C., Sharma, R. P., de Jong, R. and Stam, C. H. (1987) *Phytochemistry* 26, 2406.
50. Romo de Vivar, A., Bratoeff, E., Ontiveros, E., Lankin, D. C. and Bhacca, N. S. (1980) *Phytochemistry* 19, 1795.
51. Main, Peter "Mulan 78. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data". Department of Physics, University of York, York, England.